

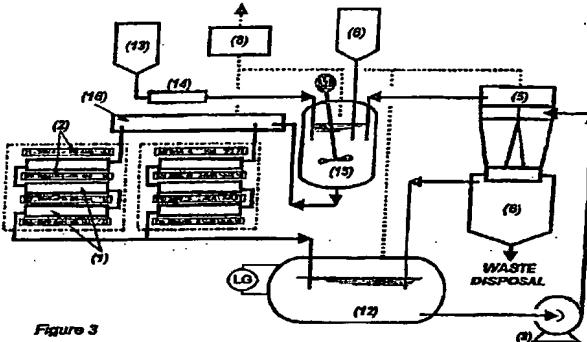
UK Patent Application GB 2 391 785 A

(43) Date of A Publication 11.02.2004

(21) Application No:	0318364.7	(51) INT CL ⁷ : A61L 2/08 2/10
(22) Date of Filing:	06.08.2003	(52) UK CL (Edition W): H5R RBU
(30) Priority Data:		(56) Documents Cited:
(31) 0218314	(32) 07.08.2002	GB 2332350 A GB 2252707 A EP 0105814 A2 WO 1992/019284 A1 GB 2309876 A GB 2165155 A WO 1999/010273 A1 JP 050305125 A
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(54) Abstract Title: Treatment of chemical and biological hazards using UV light an a titanium oxide catalyst

(57) The present invention relates to a method for the deactivation and / or destruction of hazardous materials such as chemical or biological agents. The invention further relates to apparatus for treating hazardous material and for decontaminating items that may have come into contact with it, the apparatus comprising a treatment vessel or chamber and a light source capable of irradiating a catalyst within the treatment vessel or chamber with a predetermined wavelength of light. The vessel is accessible by the user and the light source is an ultra-violet lamp for sterilising biological material. Additionally, a titanium oxide catalyst may also be used. The hazardous materials are placed in trays in which a turbulent flow, within an aqueous carrier is induced.



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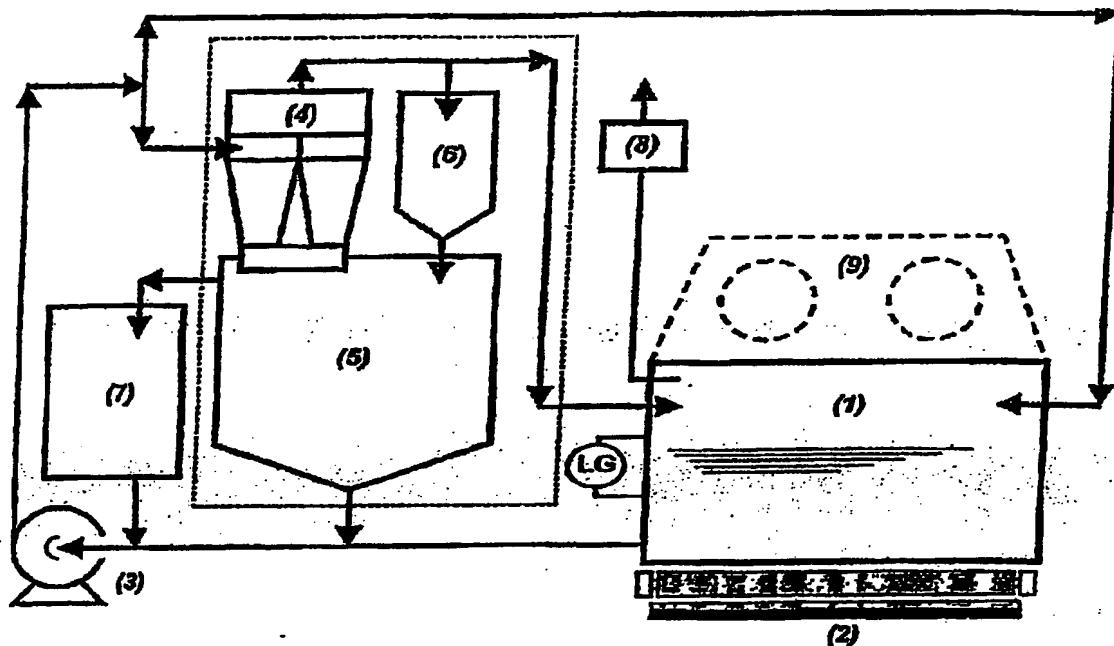


Figure 1

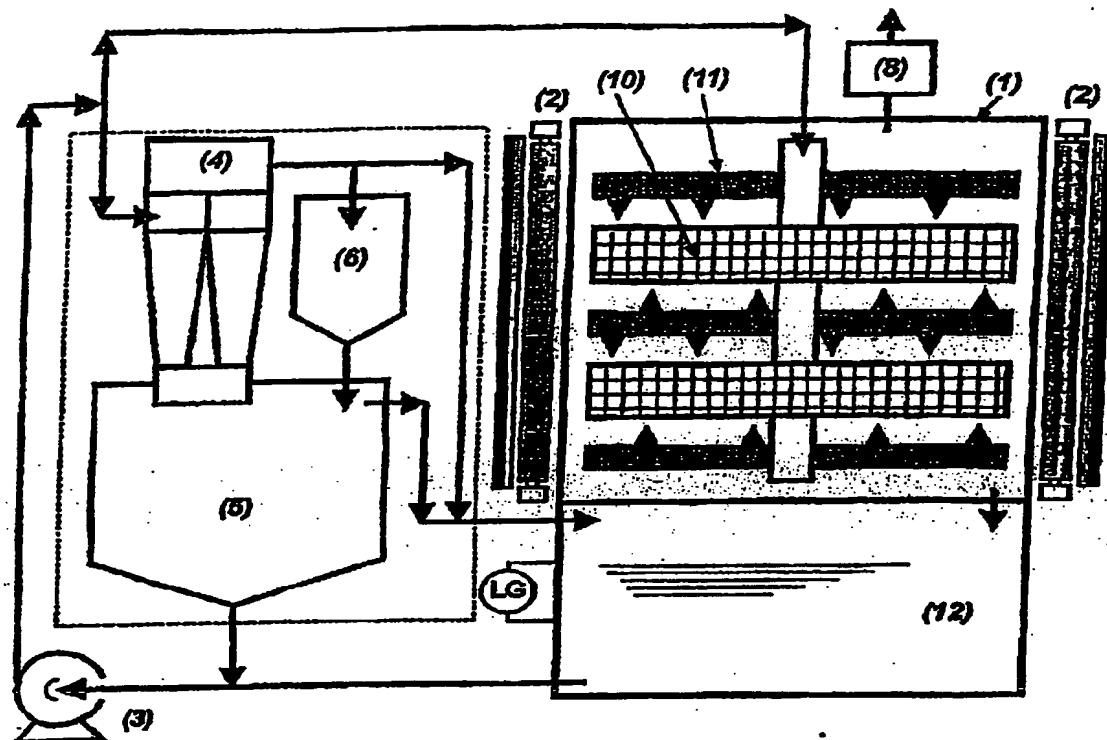


Figure 2

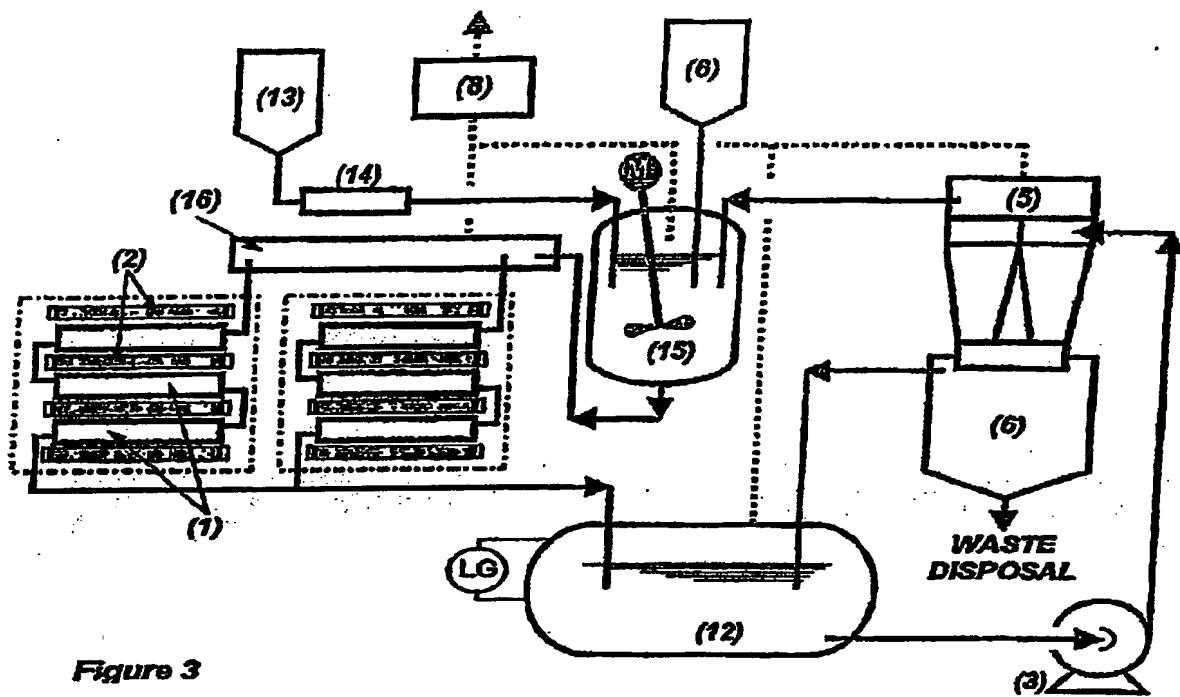
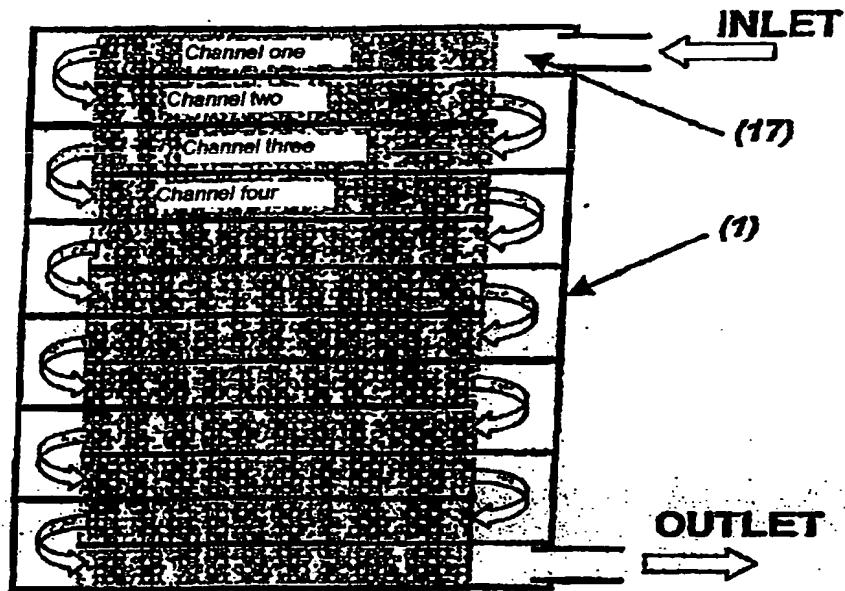
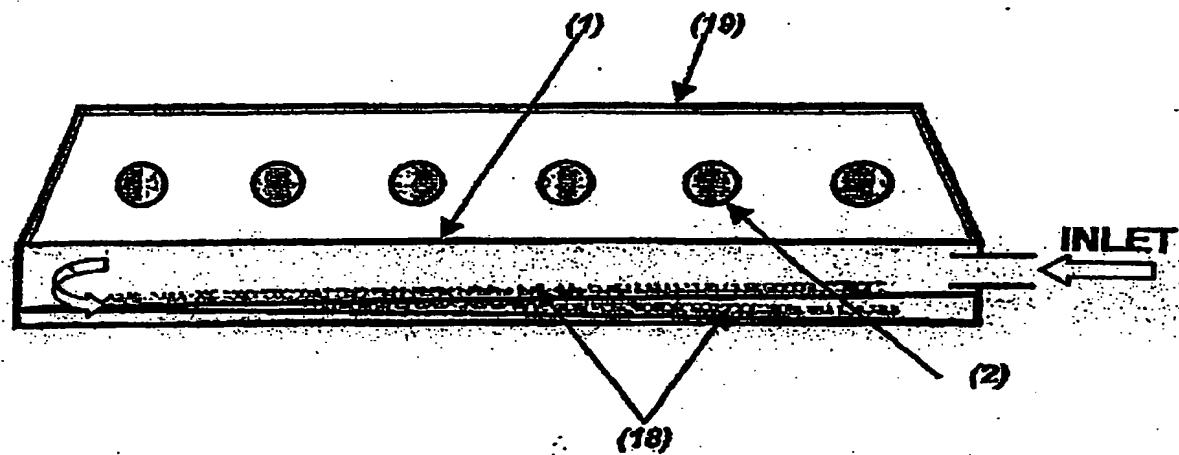


Figure 3

**Figure 4**

(1)

**Figure 5**

1 Apparatus and method for treatment of chemical and
2 biological hazards
3
4 The present invention relates to an apparatus for the
5 treatment of hazardous materials specifically prions,
6 chemical and biological agents. The invention further
7 relates to a method for using such an apparatus.
8
9 The risks associated with contamination caused by
10 chemical and biological agents of various kinds are
11 well known. Medical equipment and surgical instruments
12 are required to be sterilised to eliminate a growing
13 range of infectious agents including more recently
14 prions implicated in new variant Creusfeld Jacob
15 Disease (nvCJD). Proteins exhibit huge variation in
16 structure. However, they are formed in similar ways
17 and thus display certain structural elements and
18 characteristics that are common. The primary structure

1 of proteins is determined by the amino acid sequence
2 and pendant side groups. The amino acid chains are
3 then folded to form various secondary structures
4 designated as α -helices or β -sheets. Secondary
5 structure is determined by the folding of the amino
6 acid chains and interactions between the various side
7 groups. Further associations may also form, depending
8 on the protein's environment. For example different
9 hydrophilic and hydrophobic groups or areas within the
10 protein molecule are sensitive to the medium in which
11 the molecule may be suspended. The prion protein plays
12 an essential role in the pathogenesis of a group of
13 sporadic, genetically determined and infectious fatal
14 degenerative diseases, referred to as prion diseases,
15 or transmissible encephalopathys (TSE's), affecting the
16 central nervous system of humans and other mammals.
17 The cellular prion protein is encoded with a single
18 copy gene, highly conserved across mammalian species.
19 In prion diseases this protein undergoes conformational
20 changes involving a shift from α -helix to β -sheet
21 structure. The structures of the proteins, both native
22 and rogue, have been extensively investigated. The one
23 of most interest and immediate impact to humans is the
24 protein associated with nvCJD. What is unusual about
25 the protein that is associated with TSEs is the extreme
26 robustness it exhibits. This is thought to be due its
27 β -sheet structure. Prions are known to survive
28 temperatures in excess of 300 °C. Such proteins thus

1 represent present particular problems in terms of their
2 transmission and destruction. The nvCJD prion is known
3 to have a high affinity for stainless steel and other
4 metals posing significant difficulties for the
5 sterilisation of medical equipment, such as surgical
6 instruments. At the same time, considering hazards
7 unrelated to the medical field, chemical and biological
8 agents, such as those used as weapon materials, pose
9 significant handling and disposal risks.

10
11 For the purposes of the present application, the term
12 "hazardous material" means any organic material that
13 may be inimical to human well being and as such may be
14 classed as a chemical or biological hazard. "Hazardous
15 material" includes, but is not restricted to, viral
16 material, bacterial material, prions, proteins, lipids,
17 chemical and biological agents / material with
18 associated organophosphate bases, organic waste or by-
19 products associated with pharmaceutical processes and
20 blood products, and further includes all of said agents
21 in isolation and when found within, on the surface of
22 or bonded to other material, instruments or equipment.
23 The term "target material" is used throughout in
24 reference to a "hazardous material" which is to be
25 treated according to the method of the invention.

26
27 The term "treatment" is used in its broadest form and
28 encompasses the deactivation and destruction of
29 hazardous material. Relatively minor modifications to

1 the structure or conformation of a particular agent may
2 be sufficient to render it inactive without the need
3 for the agent to be destroyed or decomposed into
4 constituent elements.

5
6 While some methods for treating such agents are known,
7 these typically involve the use of reagents which are
8 themselves difficult to handle and which have
9 associated safety issues. Fluorine and ozone for
10 example may be effective in catalysing such processes,
11 but create significant handling problems and are not
12 suited to use in an open bath apparatus. Furthermore
13 some prior art processes are required to be carried out
14 at very high temperatures and / or pressures. The
15 apparatus used in such processes is necessarily complex
16 and expensive in light of the associated handling
17 difficulties.

18
19 There remains therefore a need for a method for the
20 deactivation or destruction of prions, chemical and
21 biological agents, which is effective, efficient and
22 broadly applicable. There is a particular need for an
23 apparatus and a treatment method that can be used to
24 sterilise or decontaminate equipment and instruments
25 that may have come into contact with hazardous
26 material. The present invention as set out below
27 provides such an apparatus and a method for its use.

28

1 Accordingly, in a first aspect the present invention
2 provides apparatus for treating hazardous material and
3 for decontaminating items that may have come into
4 contact with such material. In its broadest form such
5 apparatus comprises an operator accessible treatment
6 vessel or chamber and a light source capable of
7 irradiating a catalyst within the treatment vessel or
8 chamber with a predetermined wavelength.

9

10 A first embodiment of the invention provides an
11 apparatus, for batch treatment of hazardous material,
12 comprising a treatment vessel for holding material to
13 be treated; a light source for irradiating the contents
14 of the treatment vessel; circulation or agitation means
15 and progress and / or by-product monitoring means. The
16 treatment vessel may comprise a 'glove box' type lid
17 facilitating manipulation of the bath contents by an
18 operator. An automatic light source cut-off may be
19 incorporated in order to enhance operator safety.

20

21 A second embodiment provides an apparatus comprising a
22 treatment vessel having one or more decontamination
23 trays for holding hazardous material or items to be
24 treated, a light source for irradiating the contents of
25 the treatment vessel, medium distribution means for
26 circulating a carrier medium within and / or through
27 the apparatus and by-product monitoring means.

28

1 A third embodiment provides an apparatus comprising a
2 holding tank for holding a carrier medium; a catalyst
3 hopper for holding a catalyst; a mixing vessel for
4 mixing the carrier medium and the catalyst; one or more
5 treatment chambers each having a housing which contains
6 a plurality of treatment beds and a light source; and a
7 distribution header for controlling the flow of carrier
8 medium and catalyst into the treatment chambers.
9 Preferably, each treatment bed comprises means for
10 inducing turbulent flow within the carrier medium
11 flowing therein.

12
13 A second aspect of the present invention provides a
14 method for the deactivation and / or destruction of
15 hazardous material comprising the step of irradiating
16 the hazardous material in the presence of a catalyst
17 with light having a wavelength in the range of 310 nm
18 to 400 nm. The method of the invention causes
19 sufficient chemical modification of the hazardous
20 material so as to deactivate or destroy it.

21
22 Preferably, the catalyst is TiO_2 in either rutile or
23 anatase form and preferably the method is carried out
24 at ambient temperature (of between about 15 to 35 °C)
25 and pressure (of between about 1 to 5 bar).

26
27 The method may be carried out in any water based
28 carrier medium that is compatible with the target
29 material and catalyst. Preferably the carrier medium

1 is water. Judicious choice of treatment medium is
2 required in order to ensure reliable and effective
3 treatment. In particular when considering the
4 treatment of objects or instruments contaminated with
5 prions for example the physical characteristics of the
6 apparatus and method should facilitate a suitable
7 reaction interface. This involves consideration of the
8 composition and viscosity of the carrier medium and the
9 path-length of the apparatus such that the target
10 material, catalyst and photons from the light source
11 are brought together in a manner suitable to effect
12 treatment. It follows that a medium that is relatively
13 low in viscosity and has appropriate optical
14 characteristics (over the wavelength(s) of the light
15 source) is desirable. In other words, the viscosity
16 must be such as to allow the bringing together of the
17 target material and the catalyst and the configuration
18 of the apparatus and the optical characteristics of the
19 medium must allow sufficient transmission of light to
20 the target / catalyst reaction site.

21
22 Thus, the present invention provides for the treatment
23 hazardous material such as prions linked with human or
24 animal nvCJD in both α and β forms and for treatment of
25 instruments and equipment that may have been
26 contaminated with said material. The method, and
27 apparatus for implementing it, are also applicable to
28 the destruction of chemical agent material, typically

1 organophosphate based systems, as typified by VX or
2 Sarin, but additionally blistering and choking agents
3 as typified by Mustard Gas and Tear Gas. Depending
4 upon the conditions employed, the invention provides
5 for total destruction of some hazardous material by
6 breaking it down into its constituent parts,
7 principally carbon dioxide, nitrogen, water and
8 inorganic salts, or alternatively provides for
9 sufficient modification of target materials so as to
10 render them inactive. The invention can also
11 deactivate or destroy many other biohazards, viral and
12 bacteriological material, and many commonly
13 industrially produced organic materials. Furthermore,
14 the method of the invention can be employed to
15 decontaminate materials, equipment, instruments and the
16 like which may have come into contact with hazardous
17 material.

18
19 The method of the invention represents an efficient
20 means of deactivating and / or destroying of hazardous
21 material under mild conditions on a batch basis.
22 Further advantages of the invention are described
23 below.

24
25 The various aspects of the invention are described in
26 detail below with reference to the accompanying
27 drawings in which:

28

1 Figure 1 shows a first embodiment of an apparatus
2 according to the invention;
3 Figure 2 shows a second embodiment of an apparatus
4 according to the invention;
5 Figure 3 shows a third embodiment of an apparatus
6 according to the invention; and
7 Figures 4 and 5 are more detailed views of the
8 treatment chamber of the embodiment shown in Figure 3.

9

10 In the drawings similar reference numerals have been
11 used to designate components common to each of the
12 alternative embodiments.

13

14 In its broadest form the invention provides a
15 decontamination method for the treatment of hazardous
16 material comprising the step of irradiating the
17 hazardous material in the presence of a catalyst, with
18 light of a suitable wavelength, to deactivate or
19 destroy the target material through photocatalytic
20 oxidative processes. In general terms, the apparatus
21 of the present invention comprises (i) a treatment
22 chamber in which the catalyst and the target material
23 may be irradiated with light of a suitable wavelength
24 (and energy) and (ii) a light source capable of
25 producing the desired wavelength. The light source
26 wavelength and intensity may be adjusted to optimise
27 the process depending upon the nature of the target
28 material and the choice of catalyst. A liquid carrier,
29 preferably a water based medium, is used to introduce

1 hazardous material into the treatment chamber for
2 irradiation.

3

4 Without being bound by theory, the invention is
5 considered to be the result of an interaction of light
6 energy (photons), the catalyst and water elements that
7 forms hydroxyl radicals which cleave sections of, or
8 links in, molecules of the target material ('primary
9 effects'). The action of UV light contributes directly
10 to the breakdown of target materials through photolysis
11 of molecules present. In conjunction with the
12 formation of hydroxyl radicals hydrogen peroxide (H_2O_2)
13 is also produced. This oxidising agent assists and
14 speeds the decontamination process cycle. The primary
15 effects of hydroxyl radicals allow secondary processes
16 (such as attack by H_2O_2) to act upon vulnerable parts of
17 the molecules. The ultimate result is the break down
18 of hazardous material into simple (safe) moieties,
19 formation of inorganic salts within the carrier medium
20 and production of off-gases, such as CO_2 .

21

22 The method of the invention employing highly reactive
23 hydroxyl radicals and H_2O_2 produced through irradiation
24 of a suitable catalyst can be utilised to oxidise prion
25 proteins decomposing them to NO_x , CO_2 , water and various
26 inorganic salts. Attack on a prion protein molecule by
27 a hydroxyl radical causes selective breakage of
28 multiple bond linkages, thus permanently altering the
29 crucial relationship between amino acid units and

1 inducing changes to their proper attachment and
2 alignment to each other (and to associated components
3 such as carbohydrates and possibly lipids). This
4 effect changes the spatial configuration of the prion
5 protein impacting upon its ability to reproduce
6 properly. It is possible that even small alterations
7 in the protein composition and / or configuration are
8 sufficient to impede biological activity of a prion
9 molecule. Any alteration in the structural make-up and
10 configuration reduces the resistance of the prion to
11 further oxidative processes, such as attack by H_2O_2 ,
12 thus increasing the rate of complete oxidation of the
13 molecule.

14

15 Contact between the hydroxyl radical / hydrogen
16 peroxide production interface and the target material
17 on the equipment / instruments or the like, using the
18 water based carrier medium with the catalyst, is
19 maximised. This may be addressed by ensuring that the
20 catalyst within the water carrier is migrated to the
21 interface using suitable circulation or entraining
22 processes. Minimising the spatial offset in this
23 manner increases the effects of the short-lived
24 radicals produced upon irradiation. Spatial offset
25 distance is further aided through the use of small
26 catalytic particulate (3 - 5 microns).

27

1 Prior cleaning of gross material make take place within
2 the decontamination train, that minimises the volume of
3 material to be decontaminated, and improves throughput.

4

5 Increasing the intensity of irradiation and / or
6 increasing the surface area of catalyst irradiated can
7 increase radical production. Additional catalyst may
8 be introduced to speed the process and replace catalyst
9 extracted from the waste stream.

10 The catalyst may be any photosensitive material, which
11 allows, through illumination with light of a suitable
12 wavelength, a reaction with the associated hazardous
13 material to occur. Suitable catalyst materials include
14 for example TiO_2 , TiO_3 , ZnO , CdS , $CdSe$, SnO_2 , WO_3 , Fe_2O_3 ,
15 and Ta_2O_5 . An example of a preferred catalyst is TiO_2 .
16 Irradiation of the catalyst produces active sites (on
17 what is in effect a semiconductor surface) causing
18 water absorbed to the surface to be oxidised. Highly
19 reactive hydroxyl radicals formed in this manner react
20 with (and ultimately decompose) the target material
21 present in the system.

22

23 The catalyst may be used in any form that provides
24 suitable contact with the target material. For
25 example, the catalyst may be dispersed in the carrier
26 medium or it may be coated onto or mixed with the
27 various materials to be decontaminated or destroyed. A
28 catalyst module such as a column or tower coated with

1 catalyst material may be employed. Alternatively, the
2 catalyst may be coated onto internal surfaces of the
3 apparatus, enhancing robustness and self-cleaning
4 capability. Recovery of the catalytic material for
5 reuse, increasing efficiency of the process, may be
6 provided for as described below.

7

8 While light in the range of 310 nm to 400 nm is
9 preferred, the wavelength of light employed may vary
10 depending upon the catalyst used, the medium used and
11 the nature of the target material. The wavelength to
12 be used may be selected based on the absorption
13 characteristics of the target material, thus increasing
14 efficiency. As photo-generated hydroxyl radicals are
15 the primary agents responsible for the decontamination
16 / destruction processes various parameters may be
17 changed to optimise the effect upon any given target
18 material. The selected wavelength may be produced for
19 example using a standard mercury lamp in conjunction
20 with a suitable filter.

21

22 The method of the invention degrades target materials
23 ultimately reducing them to simple reaction products
24 such as CO₂. The evolution of CO₂ or any other reaction
25 product can thus be used to monitor the degree and rate
26 of the process. Suitably off-gas production or target
27 material break down may be monitored using techniques
28 such as Raman spectroscopy, mass spectrometry, *in vitro*

1 tests or other known techniques appropriate to any
2 particular hazardous material.

3
4 Characteristics of the method of the invention are
5 detailed in Table 1, together with comparable data for
6 various prior art methods. The 'efficiency' values
7 indicate the rate and effectiveness of electron
8 transfer during the treatment process.

Catalyst	Efficiency (eV)	Medium	Output toxicity	Temp (°C)	Pressure (bar)	Power
TiO ₂ (present invention)	3.34	Water	Very low	<36	<10	Low
Ag (II)*	1.98	Nitric acid	High	~90	10	High
Ruthenium*	1.8	H ₂ SO ₄	High	~90	10	High
Chlorination*	1.3	Water	High	~40	<10	Low
H ₂ O ₂ **	2.00	Water	Low	<36	<10	Low

1

2 Table 1. *Indicates prior art process; **Hydrogen
 3 peroxide not a catalyst as such - included
 4 for comparison purposes only.

5

6 Prior art methods (other than those detailed in Table
 7 1) include hydrogenation and methods employing molten
 8 metals or supercritical water. These additional
 9 methods all pose significant hazards themselves due to
 10 the operating conditions required in order to be
 11 effective (for example, all three require temperatures
 12 in excess of 600 °C; and hydrogenation and
 13 supercritical water methods operate at pressures of
 14 about, or in excess of, 100 bar). Treatment with
 15 fluorine, possibly the strongest oxidising agent known,
 16 is also effective, but extremely difficult and
 17 dangerous to handle.

1 The method of the invention provides an effective and
2 efficient process for the deactivation and / or
3 destruction of hazardous material, on batch or
4 continuous basis, while overcoming the shortcomings of
5 some prior art methods in terms of operational
6 requirements and characteristics. The present
7 invention facilitates decontamination treatments to be
8 carried out under ambient temperature and pressure
9 conditions through a method and apparatus which has
10 minimal moving parts, is easy to maintain and operate
11 and which is readily scalable.

Class of Compound	Examples
Alkanes	Methane; pentane; heptane; n-dodecane; cyclohexane, paraffin
Haloalkanes	mono-, di-, tri-, and tetrachloromethane; dichloropropane Pentachloroethane; di and tribromoethane; 1,2-dichloropropane
Aliphatic Alcohols	methanol; ethanol; n- and iso-propanol; butanol; penta-1, 4-diol
Aliphatic Carboxylic Acids	methanoic, ethanoic; trichloroacetic; butyric; oxalic
Alkenes	propene; cyclohexene
Haloalkenes	di-, tri- and tetra-chloroethene; hexafluoropropene
Aromatics	benzene; naphthalene, Tributyl Phosphate
Haloaromatics	chloro and bromobenzene; chlorobenzenes; halophenols
Phenols	phenol; hydroquinone; catecol; resorcinol; cresol, nitrophenol
Aromatic Carboxylic Acids	benzoic; phthalic; salicylic
Polymers	polyethylene; PVC
Surfactants	polyethylene glycol; p-nonyl phenyl ether; sodium dodecyl benzene sulphonate; paraxon; malathion
Herbicides	methyl viologen; atrazine; simazine; bentazon
Pesticides	DDT; parathion; lindane, monocrotophos
Dyes	methylene blue; rhodamine B; methyl orange; fluorescein
Explosives	Trinitrotoluene
Cyanotoxins	Microcystins, Anatoxin-a
Bacteria	E.Coli., Serratia marcescens,
Proteins	

Table 2

1 Table 2 lists compounds successfully destroyed using
 2 the present invention. Tributyl phosphate, appearing
 3 in the 'Aromatics' class, is a simulant for nerve
 4 agents.

5

Material	Concentration (% v/v)	Wavelength (nm)	Time (min)	Efficiency (%)
Methanol	0.1	385 +/- 10	20	99.5
Paraffin	0.1	385 + / - 10	40	99.75
Benzene	0.1	380 + / - 10	60	99.9

6 Table 3.

7

8 Table 3 details a number of test materials and the
 9 conditions under which they were treated. In each case
 10 treatment was carried out at atmospheric pressure and
 11 at room temperature. The treatment efficiency (which
 12 in the case of the three test materials corresponds to
 13 destruction of the compounds in question) was measured
 14 using spectrophotometric techniques.

15

16 The specific embodiments of an apparatus according to
 17 the invention described below may each be provided with
 18 a circulation system, a catalyst feed mechanism, and a
 19 catalyst recovery system. In addition there may be a
 20 flushing mechanism to remove excess free catalyst
 21 deposits from the cleaned instruments or tools and
 22 materials prior to final removal and drying. Larger
 23 units having the same basic unit structure may be

1 complemented by material towers coated with the
2 catalyst through which the contaminated material in the
3 water-based matrix is allowed to percolate, thus
4 increasing exposure of the contaminants to the catalyst
5 and UV sources.

6
7 Prior cleaning of gross material make take place within
8 the decontamination train, that minimises the volume of
9 material to be decontaminated, and improves throughput.

10
11 A first embodiment of an apparatus according to the
12 invention is shown schematically in Figure 1. The
13 apparatus comprises a treatment chamber or bath (1), a
14 light source (2), a circulation pump (3), an off-gas
15 monitor / treatment unit (8), a catalyst recovery
16 system (4) and a holding tank (5). A catalyst hopper
17 (6) and a medium storage unit (7) for storing the
18 catalyst and carrier medium prior to use are also
19 provided. This first embodiment has been designed for
20 small quantity throughput of, for example, surgical
21 instruments for decontamination or for destruction of
22 small quantities of target material. Manual
23 manipulation of items in the treatment chamber may be
24 facilitated through use of a glove-box type lid (9).
25 This apparatus is designed for operation by medical
26 staff in for example medical or dental practices.

27
28 Catalyst material and carrier medium are introduced
29 into the holding tank (5), from the catalyst hopper (6)

1 and the medium storage unit (7) respectively, and from
2 there into the treatment chamber (1). The catalyst is
3 typically suspended in the carrier medium and suitable
4 stirring means may be provided in order to ensure that
5 suspension is maintained and that the suspension
6 circulates within the chamber (1). The contaminated
7 equipment or target material (not shown) is placed in
8 to the bath; the lid closed and interlocks (not shown)
9 engaged before the process commences. In order to
10 maintain the catalyst in suspension within the carrier
11 medium during the process, the medium is circulated
12 through the system by using suitable means. This
13 facilitates maximum irradiation of the catalyst
14 simultaneously allowing the catalyst particles to
15 contact the interface with the target material. A
16 circulating pump (3) is used for the removal of
17 catalyst via the catalyst recovery system (4) at the
18 end of the process run. The catalyst recovery system
19 (4), typically takes the form of a cyclone separator.
20 The level of catalyst in the system is monitored via
21 the process controller (not shown) and adjusted to the
22 required level. The carrier medium is circulated
23 within the bath (1) during the
24 decontamination/destruction process and may be replaced
25 or replenished from the medium storage unit (7) or via
26 the catalyst recovery system (4). The process
27 controller (not shown) is used to monitor the overall
28 process, including monitoring off-gas production within
29 the off-gas monitor/treatment system (8). The off-gas

1 monitoring system (8) provides the means by which the
2 primary process status is monitored. The destruction
3 of organic elements produces CO₂, when no further CO₂
4 production is detected the treatment process may be
5 regarded as complete. The residual CO₂ given off is
6 collected by use of an active charcoal filter fitted
7 into the off-gas system (8). Sampling can be
8 facilitated in order to allow for conformity *in vitro*
9 testing, spectroscopic analysis or the like to take
10 place. Once completion of the process has been
11 confirmed the used carrier medium can be disposed of in
12 a recognised manner and the apparatus may be flushed
13 with fresh medium. The flushing process enables all the
14 areas within the apparatus that may have been
15 contaminated by target material to be cleaned, although
16 the system is inherently self-decontaminating. The
17 carrier medium within treatment chamber (1) is then
18 topped-up prior to next usage and the medium in the
19 holding tank (7) replaced. While the method of the
20 invention may generally be carried out at, or close to,
21 atmospheric pressure, materials may be passed through
22 the apparatus under higher pressure particularly during
23 catalyst recovery and / or cleaning stages.

24

25 Access to the treatment chamber (1) for this activity
26 may be provided by a glove box lid arrangement (9).
27 This allows for function (if necessary), dismantling
28 and scrubbing of instruments or equipment to remove
29 stubborn or hidden contaminants. These are

1 subsequently circulated and destroyed in the treatment
2 chamber during the treatment process. Safety
3 interlocks may be employed to minimise any risks to
4 personnel during operation, particularly when
5 introducing target material in to the apparatus.
6 Switching means are provided for deactivating the light
7 source automatically when the bath lid (9) is opened.
8
9 Prior cleaning of gross material make take place within
10 the decontamination train, that minimises the volume of
11 material to be decontaminated, and improves throughput.
12
13 A second embodiment is shown schematically in Figure 2.
14 This apparatus is designed for use in hospitals or
15 larger clinics with high throughput of surgical
16 instruments for decontamination. It is designed for
17 operation by dedicated staff with training in the
18 decontamination of surgical instruments and equipment.
19
20 The apparatus comprises a treatment chamber (1) having
21 decontamination trays (10) an ultraviolet light source
22 (2) and a medium distribution system (11). Catalyst
23 from the catalyst hopper (6) and / or a catalyst
24 recovery system (4) are introduced into a holding tank
25 (5). The contaminated equipment or product is placed
26 in the decontamination trays (10) and the trays (10)
27 are lowered into the treatment chamber (1). The lid is
28 closed and interlocks engaged before the process is
29 allowed to start. In order to maintain the catalyst in

1 suspension within the medium, the medium is circulated
2 by means of a circulation pump (3) and a medium
3 distribution system (11) having a plurality of rotating
4 spray heads (not shown). The distribution system (11)
5 creates a pressure jet effect that develops a catalyst
6 laden mist or aerosol within the treatment chamber (1)
7 which facilitates optimum contact / interaction between
8 the UV light, catalyst and target material on the
9 contaminated instruments. The carrier medium drains to
10 the bottom of the treatment chamber (1) where it is
11 collected in a circulation header tank (12) which in
12 turn feeds the circulation pump (3). At the end of the
13 treatment process any excess catalyst is recovered from
14 the medium via a catalyst recovery system (4). As
15 described above, a process control (not shown) is
16 provided to monitor progress of the treatment by means
17 of off-gas monitor / treatment system (8). Upon
18 completion of the treatment process, the lid is
19 removed, trays raised and the decontaminated
20 instruments removed.

21

22 The medium, including suspended catalyst, may be
23 circulated directly through the treatment chamber (1)
24 from the holding tank (5) during the decontamination
25 process or via the catalyst treatment unit (4) during
26 the catalyst recovery cycle. Carrier medium is sampled
27 for conformity / quality maintenance as described in
28 relation to the previous embodiment. The medium level
29 within the circulation header tank (12) is monitored

1 prior to and during operation and is topped-up as
2 required.

3
4 The third embodiment, shown schematically in Figure 3
5 with details of the treatment chamber arrangement shown
6 in Figures 4 and 5, is designed for either high or low
7 volume destruction of high level bio-hazards such as
8 chemical or biological agent materials, prion
9 contaminated material or the like (and may be adapted
10 to handle solid, liquid or gas phase hazardous
11 materials). It is envisaged that such a system would
12 be operated in a restricted area by dedicated and
13 suitably trained staff.

14
15 The apparatus comprises a series treatment chambers (1)
16 the number and configuration of which may be adapted
17 depending upon the nature and quantity of material to
18 be treated. The target material in a suitable pre-
19 prepared state is introduced from a target material
20 hopper (13) under control of metering means (14) into a
21 mixing vessel (15). The carrier medium is fed in to
22 the mixing vessel (15) from the circulation header tank
23 (12) by the circulation pump (3) and catalyst is added
24 from a catalyst hopper (6). The pre-treatment
25 preparation of the target material may include but need
26 not be limited to the breaking down of solids into
27 smaller particles, the suspension of solid particles in
28 a liquid or the absorption of a gas into a liquid. The
29 target material, medium and catalyst mixture cascades

1 into distribution header (16) from which it enters the
2 treatment chambers (1). This method of controlling the
3 flow of the mixture removes any potential pressure
4 other than the hydrostatic head determined by the
5 relationship between the mixing vessel (15) and the
6 distribution header (16). Each treatment chamber (1)
7 comprises a housing that contains a series of tray-like
8 treatment beds and a light source (2). The treatments
9 beds are designed to maximise the time which the
10 carrier medium, catalyst and target material mixture is
11 exposed to the UV light, as well as promoting the
12 formation of turbulent flow. Typically each treatment
13 bed comprises of a series of channels (17) running back
14 and forth across the bed, each channel (17) containing
15 a textured surface (18) designed to induce turbulent
16 flow within the mixture. Control of the flow in this
17 manner prevents the catalyst and target material from
18 being shielded (as could occur in a laminar flow
19 situation) and maximises irradiation effectiveness.
20 The treatment beds are configured with a light source
21 (2), optionally shrouded with a mirror, directly
22 overhead. Each treatment bed further comprises a
23 transparent top plate, typically made from quartz or
24 some other material having suitable light transmission
25 characteristics. The treatment mixture is circulated
26 around the system until the process has been completed
27 or for a suitable duration as dictated by the operator.
28 Any suspended solids, catalyst and other waste products

1 are removed via a catalyst / waste treatment system (4)
2 for storage prior to final disposal.

3
4 Specific modifications may be introduced into the
5 carrier medium composition and flow control in order to
6 create the necessary environment for the target
7 material to be suspended within the medium. For
8 example, rotary, ultrasonic or other stirring /
9 agitation means make be incorporated into the
10 apparatus.

11
12 The process is controlled using a suitable process
13 monitoring and control system. This includes
14 monitoring the off-gas status by means of an off-gas
15 monitoring / treatment system (8). The off-gas
16 monitoring / treatment system (8) also provides a means
17 for the monitoring and collection / treatment of
18 gaseous reaction products such CO₂, NO_x, SO_x and the
19 like. In order to treat these off-gases specific
20 equipment such as scrubbers and absorbers may be
21 provided. As before suitable analytical techniques can
22 be employed to monitor the course of the treatment and
23 the content of used waste products and used carrier
24 medium.

25
26 The invention is not limited to the embodiments herein
27 described which can be varied in construction and
28 detail.

1 CLAIMS

2

3 1. Apparatus for the treatment of hazardous
4 material and decontamination of items
5 contaminated with such material comprising an
6 operator accessible treatment vessel adapted to
7 hold said hazardous material or contaminated
8 items and a light source capable of irradiating
9 contents within the treatment vessel with a
10 predetermined wavelength of light.

11

12 2. Apparatus according to Claim 1 wherein the
13 treatment vessel comprises one or more trays
14 for holding the hazardous material or
15 contaminated items, further comprising
16 distribution means for circulating a carrier
17 medium within or through the apparatus.

18

19 3. Apparatus according to either one of Claims 1
20 and 2 further comprising monitoring means.

21

22 4. Apparatus according to Claim 1 further
23 comprising a holding tank capable of holding a
24 carrier medium, a catalyst hopper capable of
25 holding a catalyst, a mixing vessel
26 facilitating mixing of the carrier medium and
27 the catalyst, wherein the treatment vessel
28 comprises one or more treatment chambers each
29 having a housing containing a plurality of
30 treatment beds and a light source, and a
31 distribution header for controlling the flow of

1 carrier medium and catalyst into the treatment
2 chambers.

3

4 5. Apparatus according to Claim 4 wherein each
5 treatment bed comprises means for inducing
6 turbulent flow within the carrier medium.

7

8 6. A method for treatment for hazardous material
9 or decontamination of items contaminated with
10 such material comprising the step of
11 irradiating said material or said items in the
12 presence of a catalyst with light having a
13 wavelength in the range of from 310 to 400
14 nanometres.

15

16 7. A method according to Claim 6 wherein the
17 catalyst is TiO_2 .

18

19 8. A method according to Claim 7 wherein the
20 catalyst is TiO_2 in either rutile or anatase
21 form.

22

23 9. A method according to any of Claims 6 to 8
24 wherein the irradiation step is carried out at
25 a temperature of between about 15°C to 35°C and
26 a pressure of between about 1 bar to 5 bar.

27

28 10. A method according to any one of Claims 6 to 9
29 wherein the irradiation step is carried out in
30 an aqueous based carrier medium.



Application No: GB 0318364.7
Claims searched: 1-5

29

Examiner: Eamonn Quirk
Date of search: 7 November 2003

Patents Act 1977 : Search Report under Section 17

Documents considered to be relevant:

Category	Relevant to claims	Identity of document and passage or figure of particular relevance
X	1 at least	WO 92/19284 A1. (Baxter) see figures 2-4
X	1 at least	GB 2 165 155 A (Perrin) see whole document
X	1 at least	GB 2 309 876 A (Pet Mate) see figures and abstract
X	1 at least	EP 0 105 814 A2 (Kosukari) see figures and abstract
X	1 at least	JP 05 0305125 A (Matsushita) see WPI abstract and figures
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